



OECD Nuclear Energy Agency

9th IEM on P&T

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Progress in Partitioning: Activities in ATALANTE

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COMPLEMENTARY SEPARATIONS : WHICH PROCESSES ?

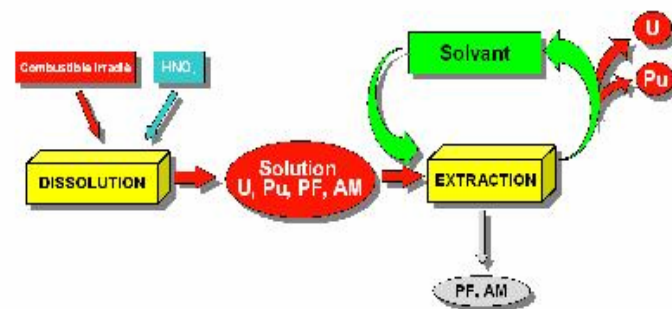


1. MAIN CRITERIA :

- *safety, efficiency, cost, secondary waste,...*
- *robust technical elements for 2006*

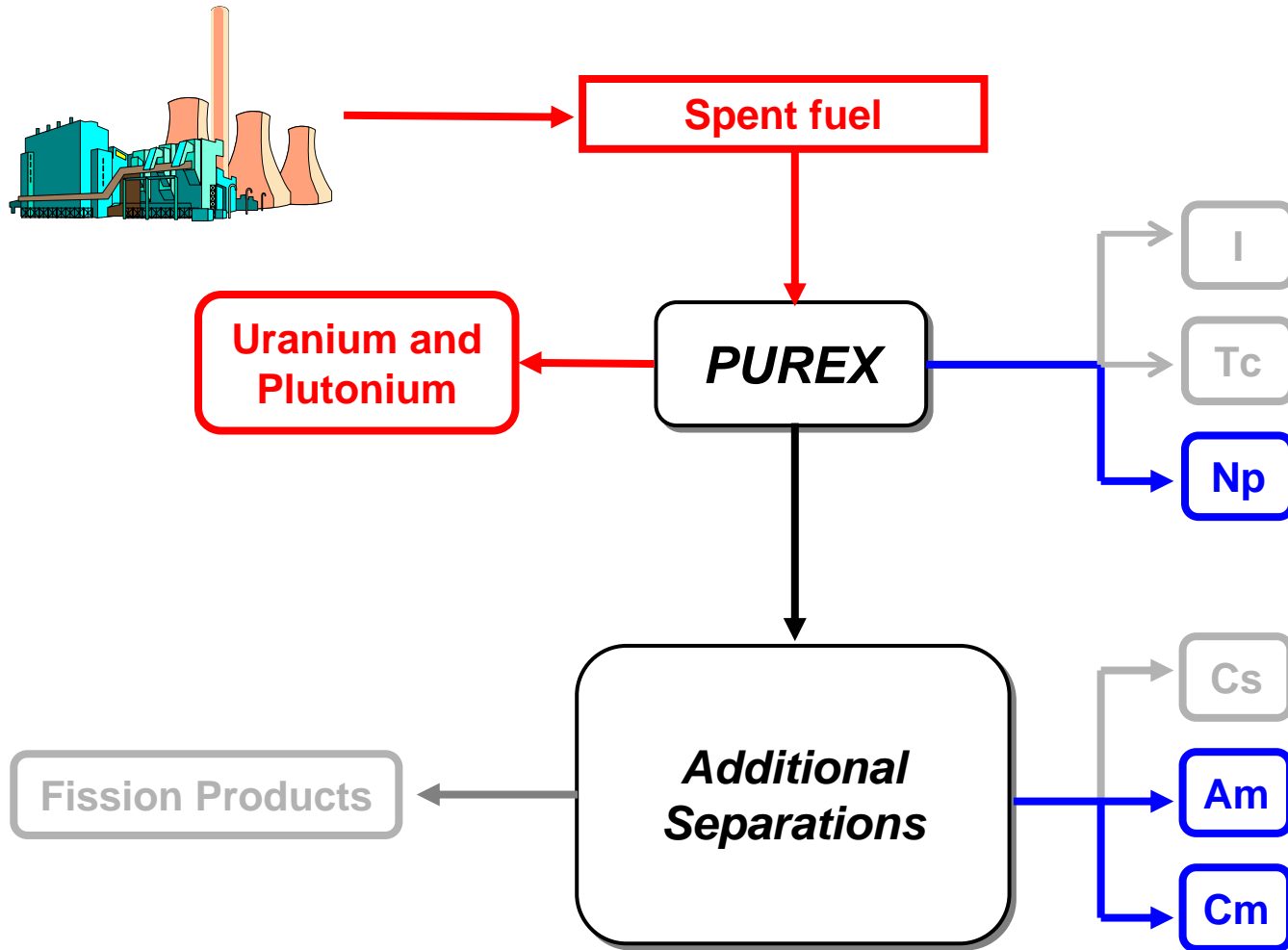
2. REFERENCE ROUTE : SOLVENT EXTRACTION

- *used for U and Pu recovery*
- *La Hague feedback :*
 - * *high separation yields*
 - * *low amount of technological waste*

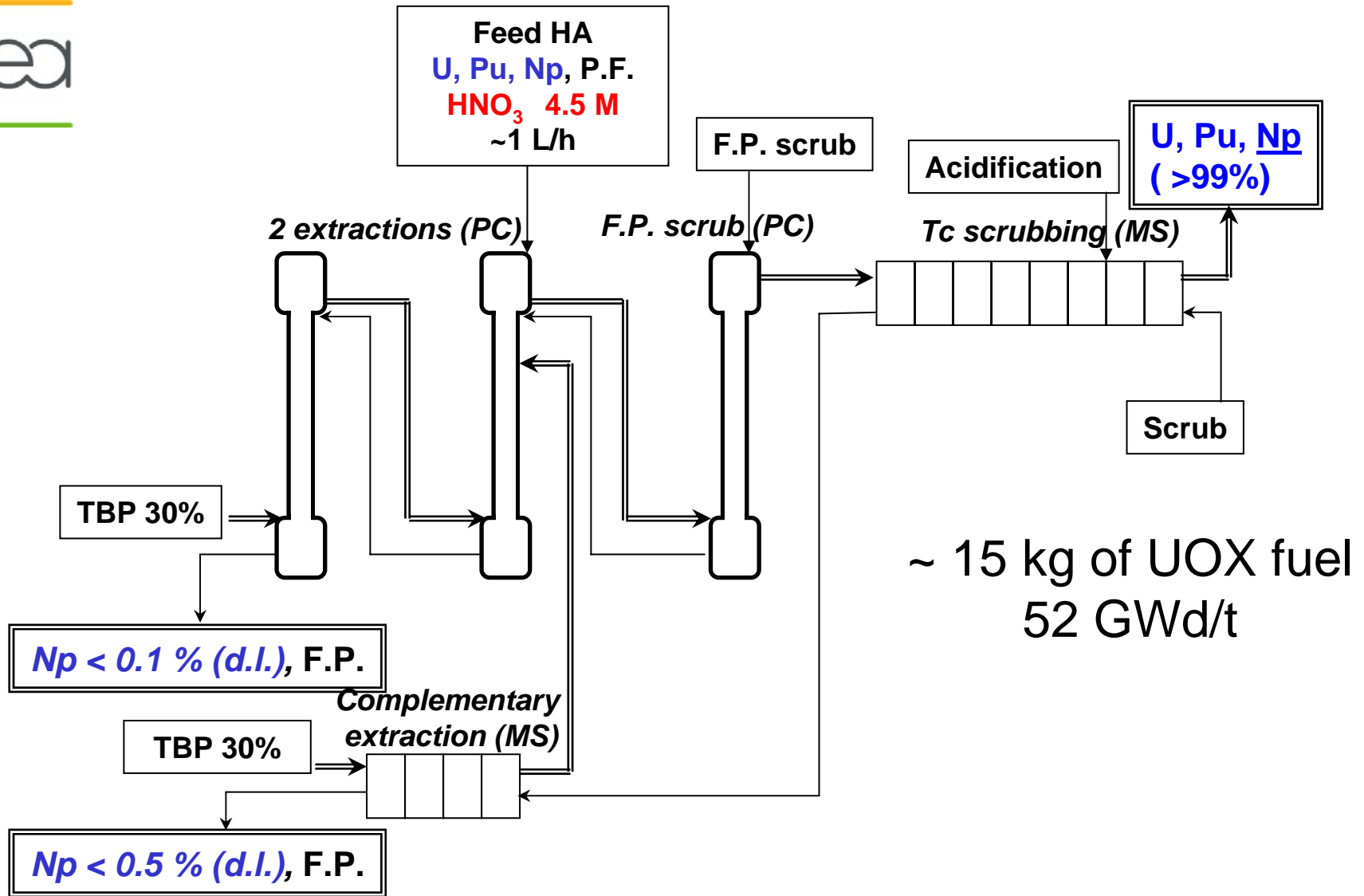


3. ALTERNATIVE ROUTE: PYRO-PROCESSES (molten salts, high temperature)

LLRNs recovery : CEA general flowsheet



Np extraction : CBP hot run, april 2005



ATALANTE « CBP » HOT CELL (commissioned 2003)

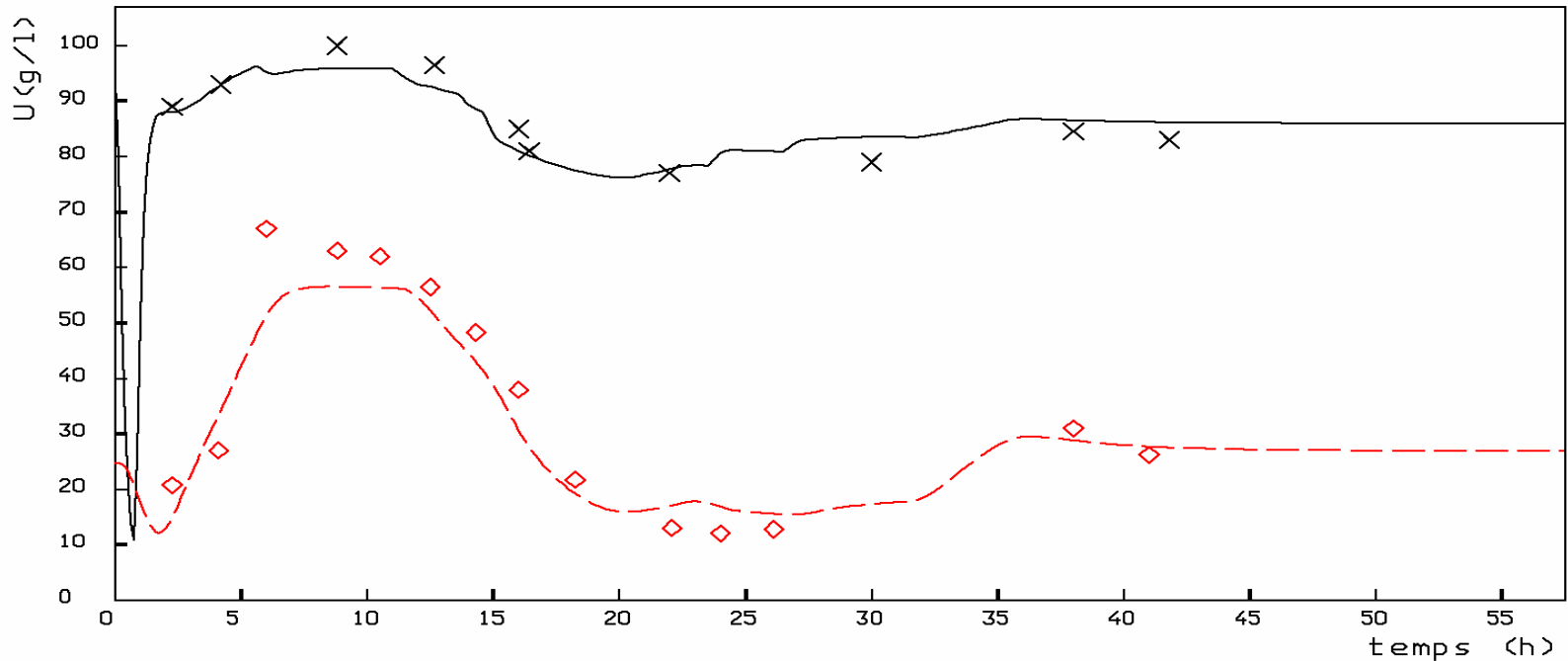


[U] in loaded solvent and scrubbing



PUREX-Np

Configuration de calcul de Parex8.3 :
config_parex_stand.cfg (9/2000)



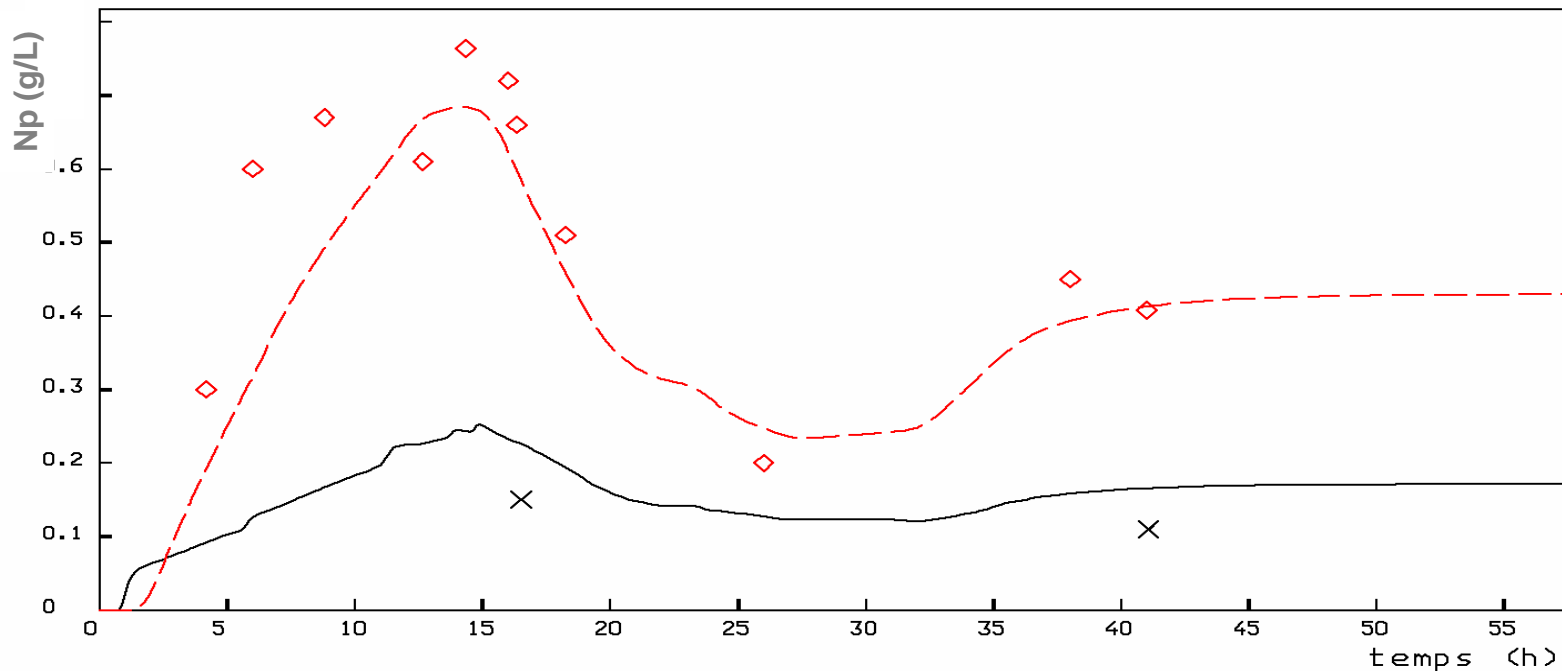
— u org. pot tampon AX1
- - - u aqu. pot tampon AS
x x u org. pot tampon AX1 exp.
◇ ◇ u aqu. pot tampon AS exp.

[Np] in loaded solvent and scrubbing

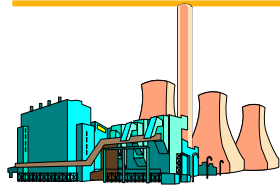


PUREX -Np

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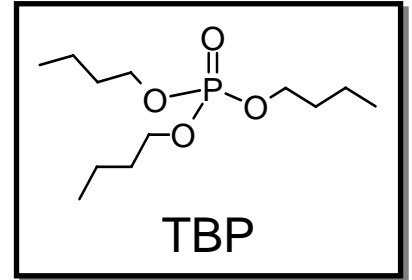
ACTINIDE SEPARATION STRATEGY



Spent Fuel

PUREX

U, Pu, Np



Two step strategy

Fission Products

DIAMEX

CO-EXTRACTION of MAs and Ln

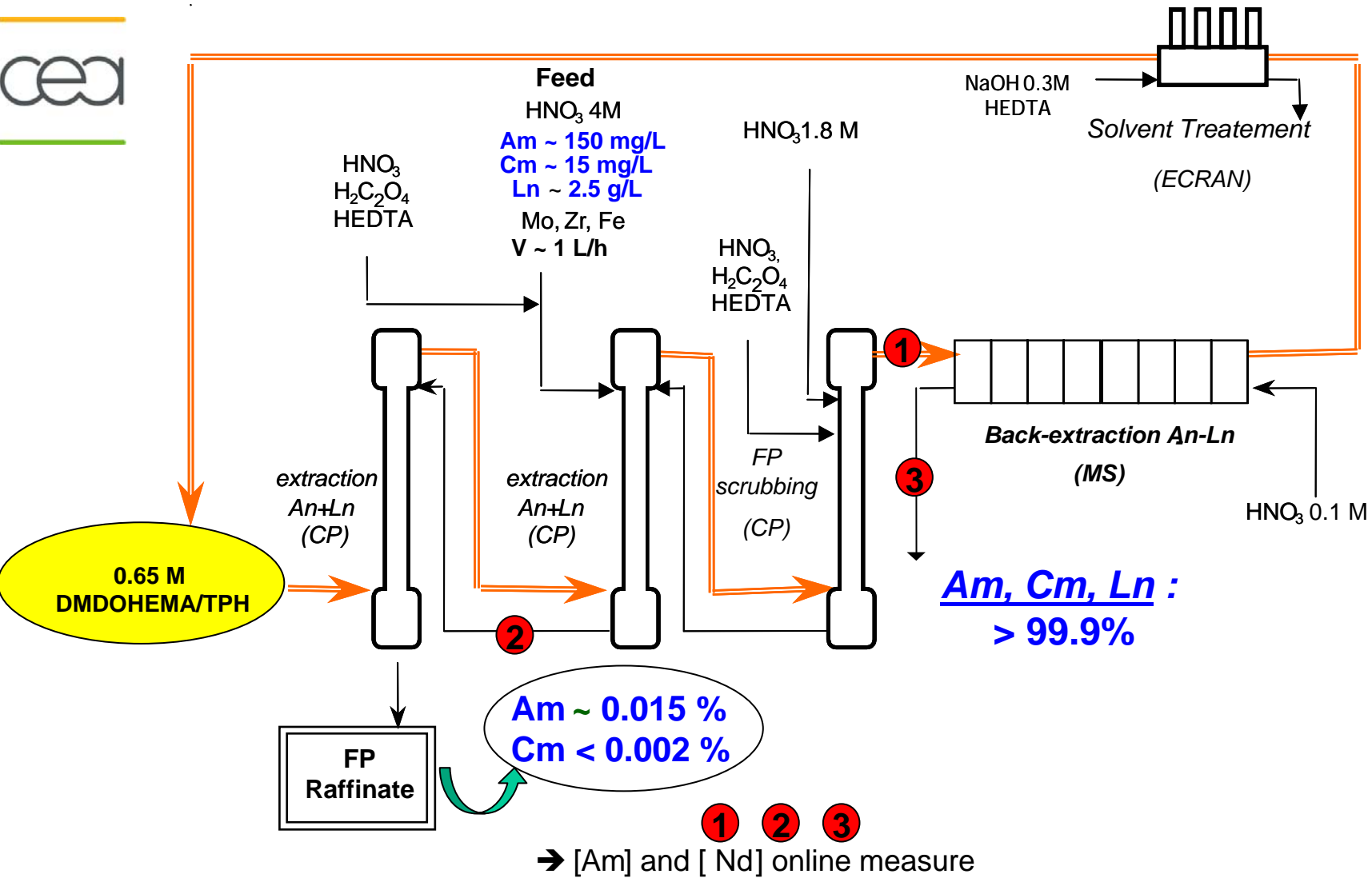
Ln

SANEX

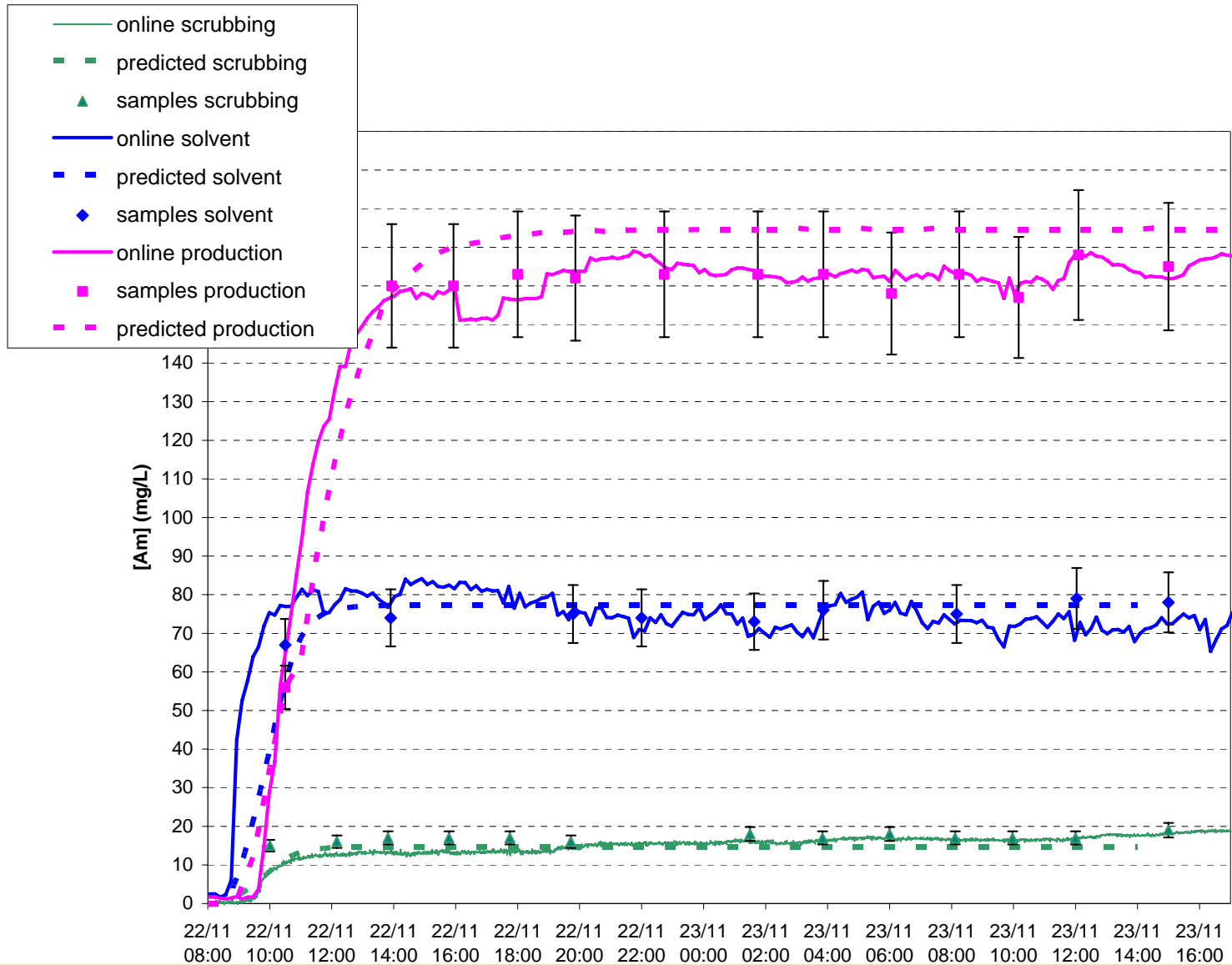
Am, Cm

SEPARATION of Mas from Ln

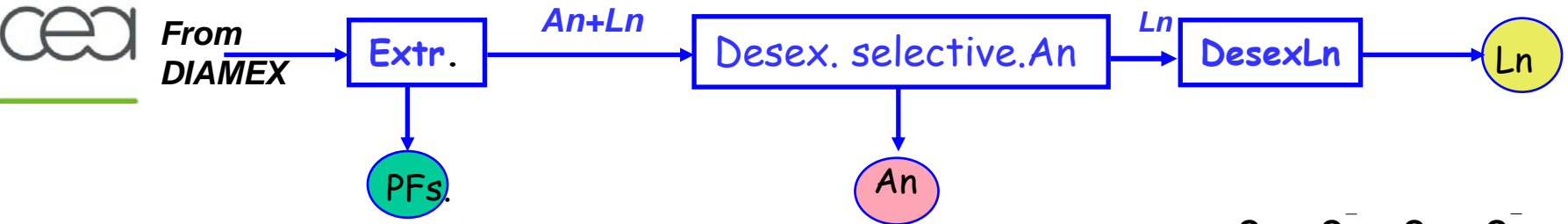
DIAMEX, « CBP » hot run, November 2005



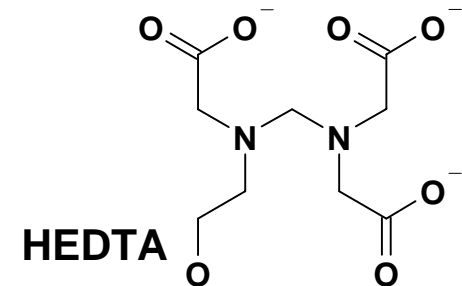
[Am] on line measure



The second step : An/Ln Separation

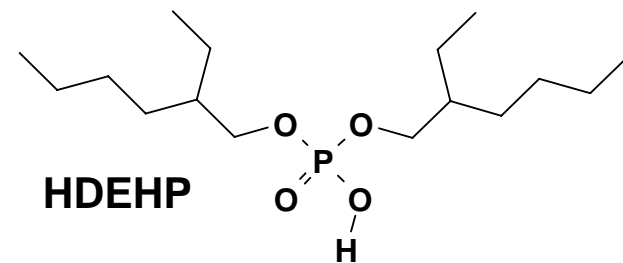
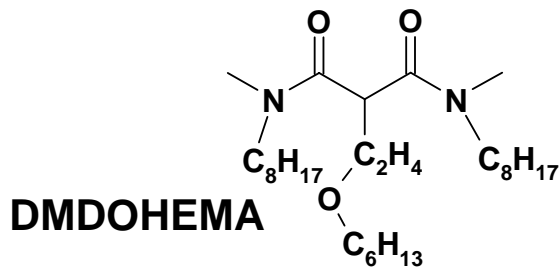


The key : an *An-selective complexant*



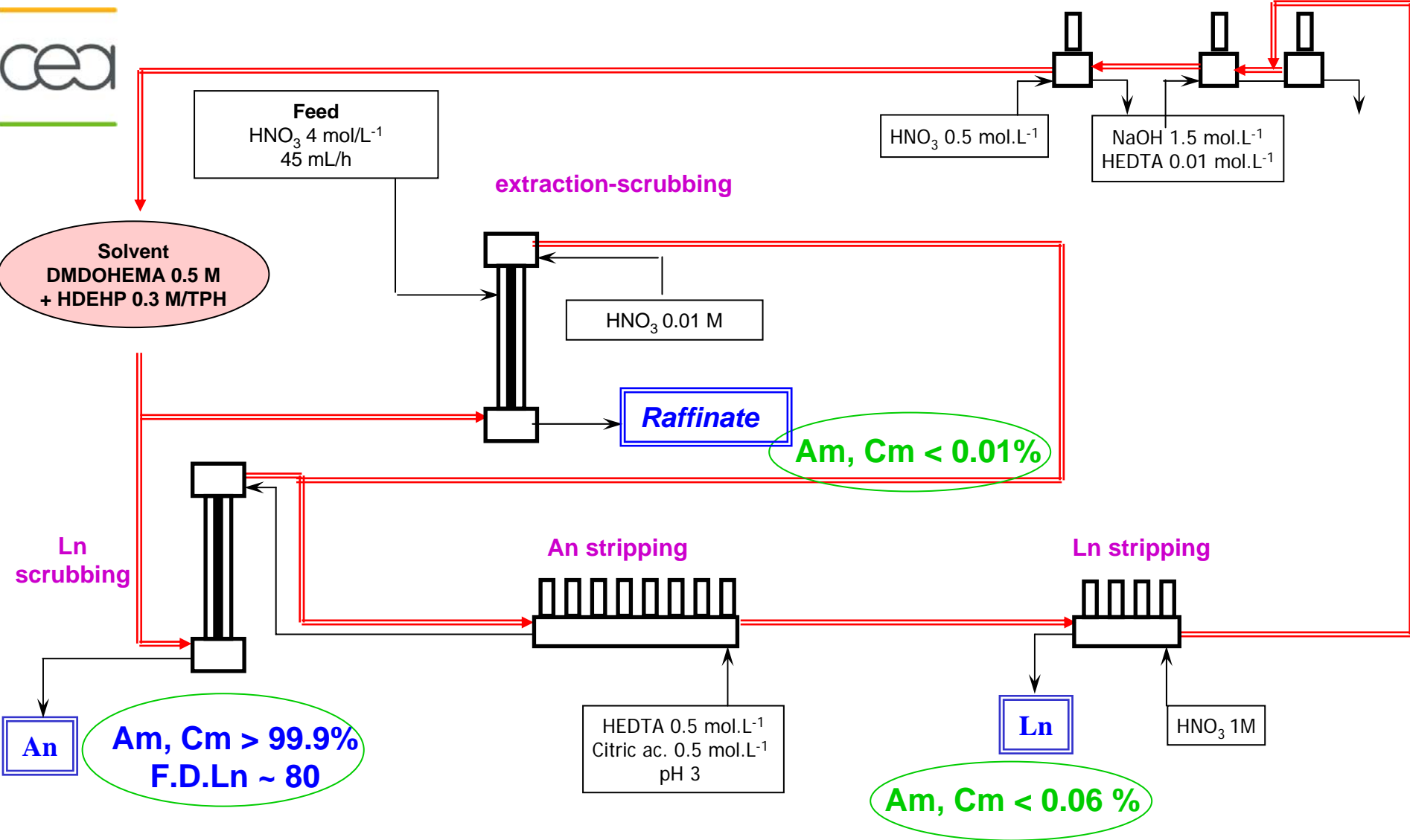
The difficulty : efficient only at low acid concentration

The solution proposed : an appropriate extractant *system*



...and towards a single-cycle concept !

SANEX Process, hot run, December 2005



Minor Actinide : *Main results*



- **Neptunium :**

⇒ *recovery yield : > 99 % (adaptation of PUREX process)*

- **Américium et curium :**

⇒ *recovery yield : 99,9 % Am + Cm*

HOW TO GO FURTHER ?

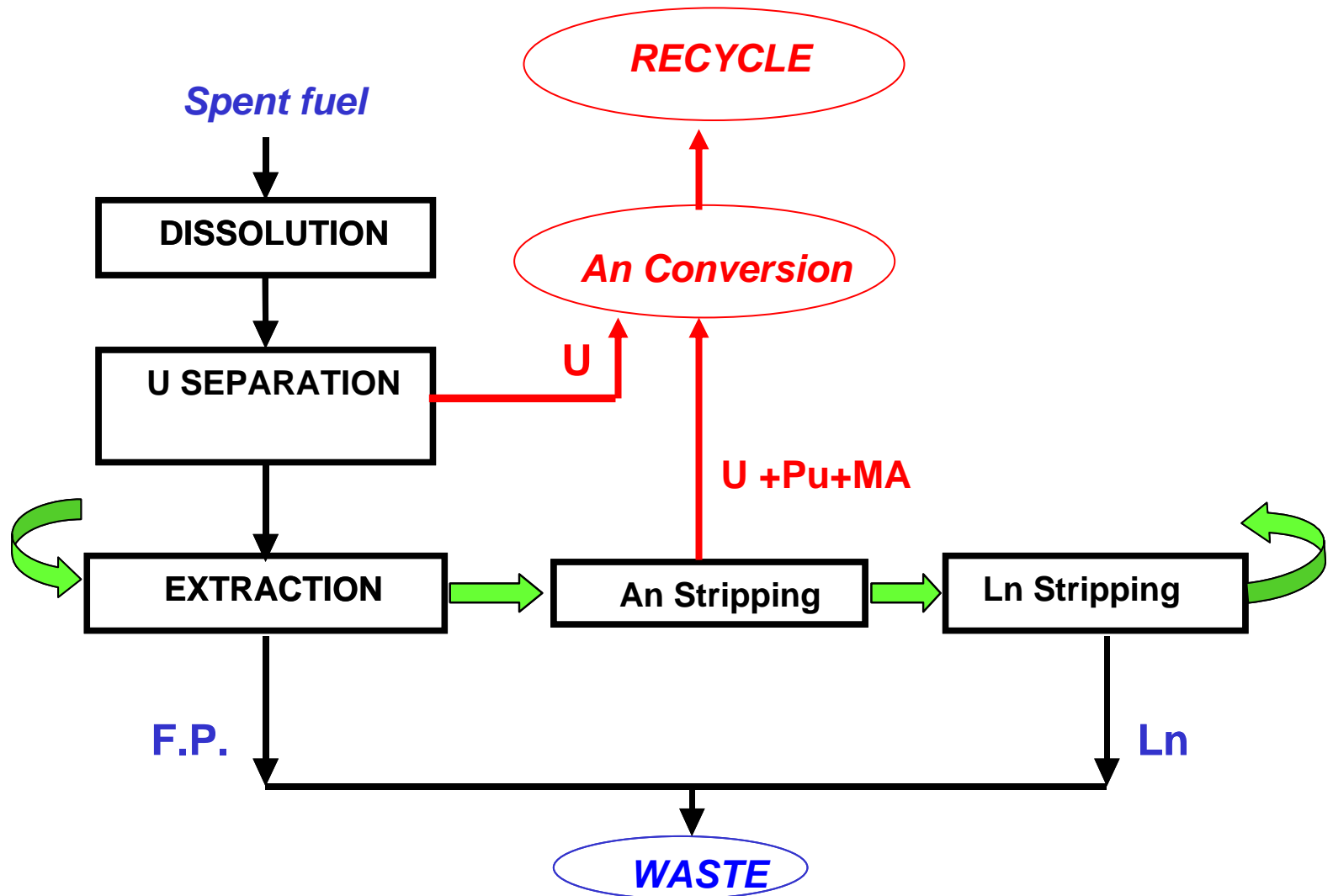


- **DIAMEX-SANEX Optimization : single cycle !**

- ***ALL-ACTINIDE GROUPED EXTRACTION***
 - *adapted DIAMEX-SANEX (with U, Pu,Np...)*

 - *other new molecules...*

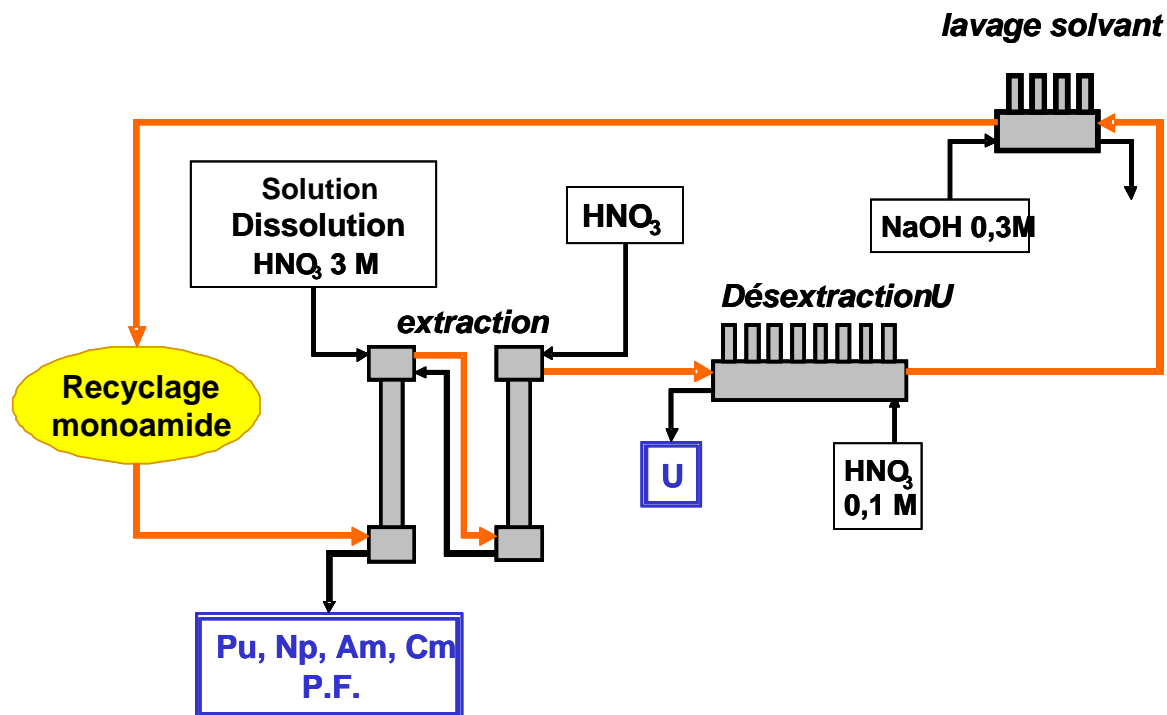
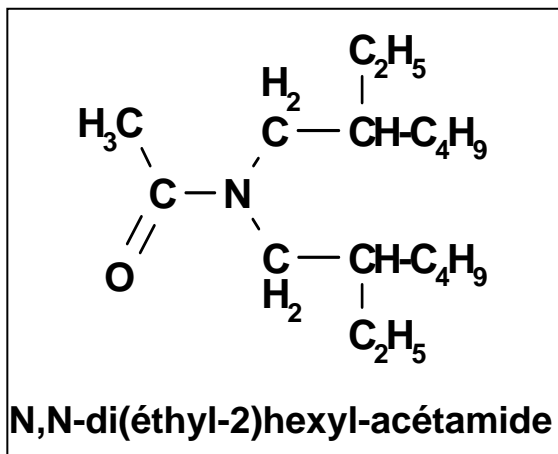
The GANEX Concept



Preliminary Uranium extraction



- ***U(VI) / other actinides separation, based on selective mono-amide extraction of uranium.***



«*Poly-actinide* » oxalate coprecipitation



1999 : U 70 % - Pu 30 %

2003 : U, Pu, Np

2005 : U, Np, Pu, Am

U 78 % - Np 1% - Pu 20 % - Am 1 %

ACTINIDE RECOVERY BY SALT/METAL EXTRACTION : ATALANTE HOT RUNS, 2004



Assessment of the Pu-Am co- extraction & separation from F.Ps by using salt/metal extraction in LiF-AlF₃ / Al(Cu) at 830°C



*High Temperature Salt/Metal
Contactor in glove-box (Atalante)*

Run 2 : Extraction (Al)

Salt before extraction: LiF-AlF₃ (85-
19.5 mole %) + **PuF₃ (11,1 wt%) +**
AmF₃ (0,12 wt%) + CeF₃ (2,50 wt%)
+ SmF₃ (0,52 wt%) + EuF₃ (0,51 wt%)
+ LaF₃ (0,52 wt%)

Alloy/Salt ratio = 1

Alloy ~ 15-17 g



Salt before extraction



Salt after extraction

en résumé....

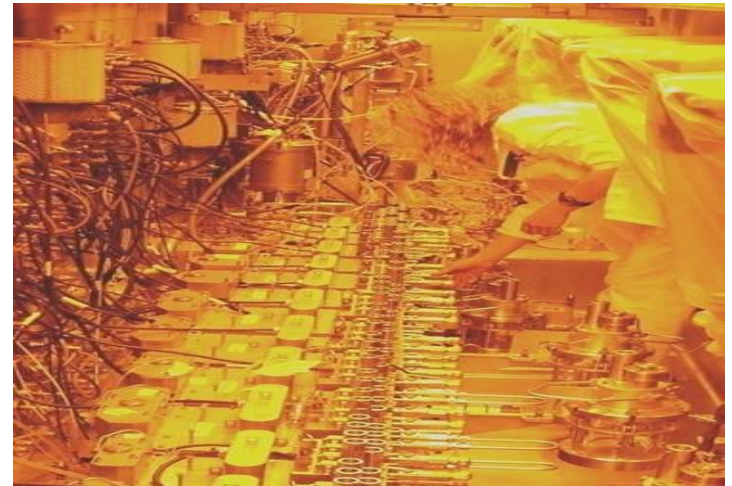
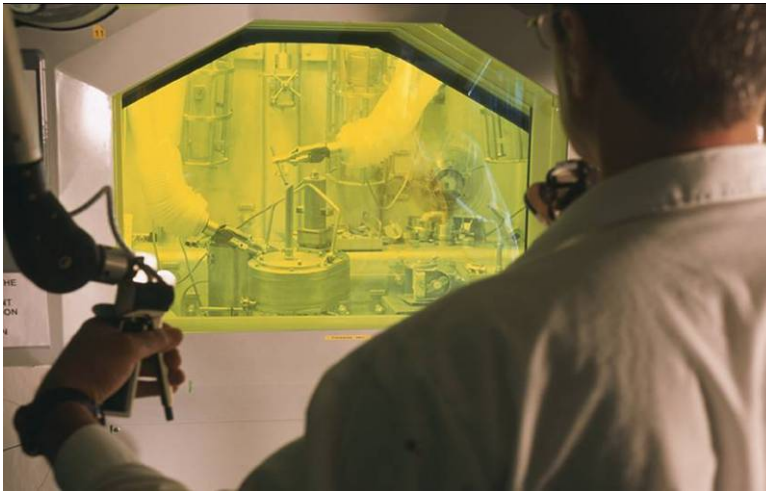


- actinides, the main target !
- Solvent extraction, a *powerfull, clean, mature technology* !
 - Many options studied for An recovery :
molecules and extraction processes have been experimented at lab-scale and « large-lab » scale.
 - co-precipitation, a very promising route to get **An solid compounds**
- Pyro-processesses : an attractive alternative !
 - CEA launched a R&D program
 - **Encouraging results**, to be pursued (technological issues!)

The ATALANTE facility...



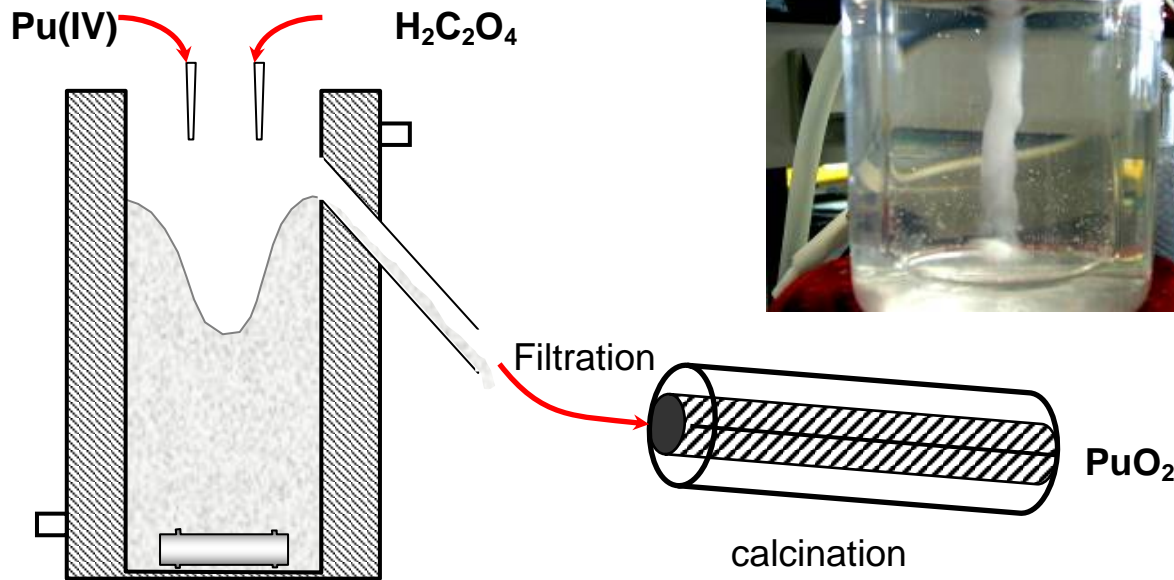
***From basic studies to demonstrative experiments ...
... up to kgs of spent fuel***



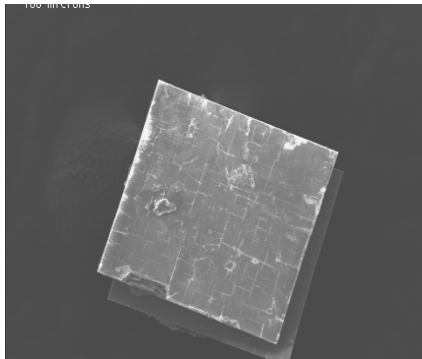
Pu industrial conversion : *oxalic precipitation*



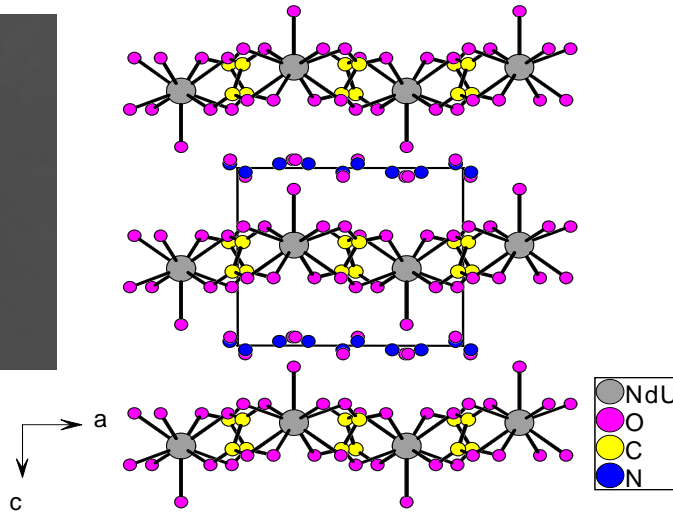
- **2 steps :**
 - $\text{Pu(IV)(C}_2\text{O}_4)_2 \cdot 6\text{H}_2\text{O}$ precipitation
 - Calcination to PuO_2
- **Operated in La Hague plant (R4/T4)**



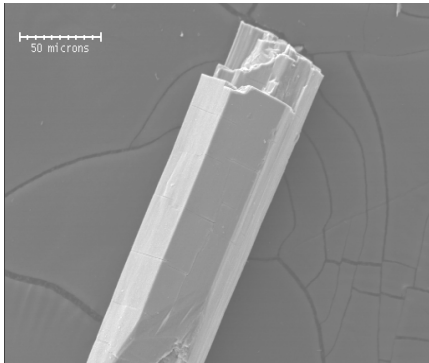
All-actinide coprecipitation ?



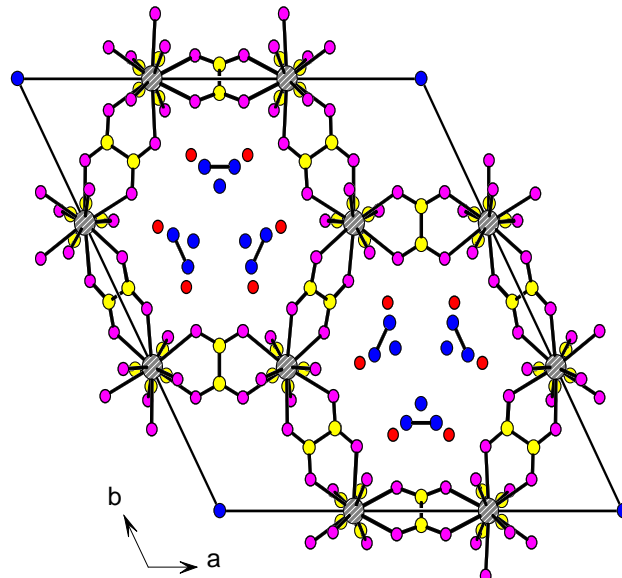
Quadratic



Same crystallographic site
For An(IV) and An(III)



Hexagonal



and solid solutions for
oxides !

PYROPROCESSES : THE ALTERNATIVE ROUTE



Frame :

- *launched in the frame of the french legislative act, Dec. 1991, on the management of nuclear waste*
- *since 2003, extended to Gen IV cycle studies*

- Very preliminary goal :

- all actinide quantitative recovery (> 99 %)
- *with a “high enough” purity level*
- *both electro-recovery and molten metals extraction*

ACTINIDE RECOVERY BY SALT/METAL EXTRACTION : MAIN RESULTS

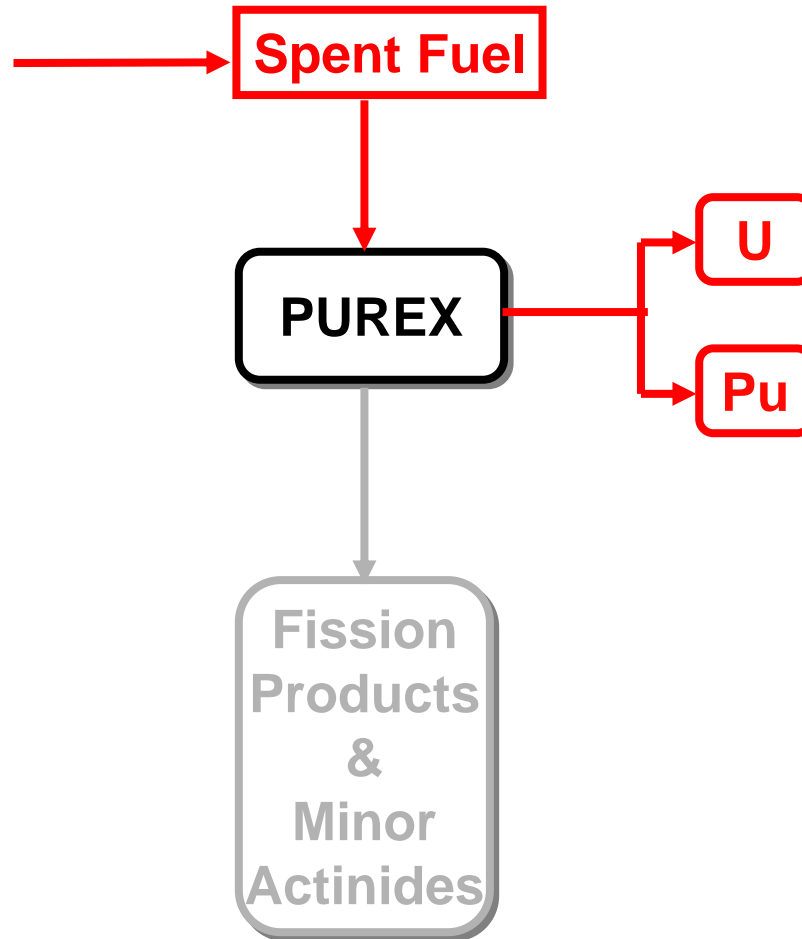
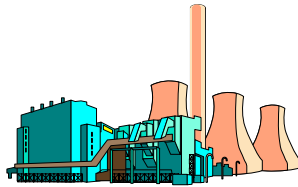


> 98 wt% of Pu and Am have been transferred to the Al-based alloy after one contact

Pure Al		
Element M	D_M	$S_{Am/M}$
Pu	273 ± 126	$0,78 \pm 0,47$
Am	213 ± 30	1
Ce	$0,162 \pm 0,02$	1315 ± 289
Sm	$0,043 \pm 0,004$	4954 ± 1139
Eu	$< 0,029$	> 7345
La	0,03	7100

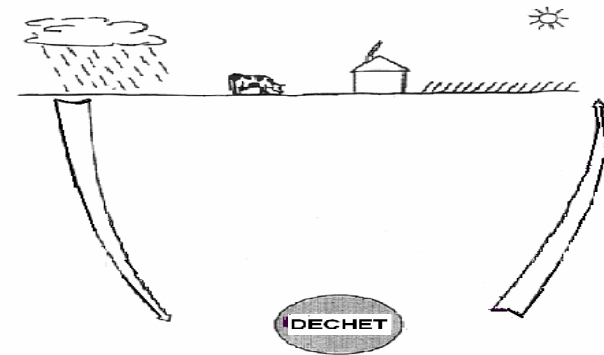
$D_{Cm} = 185 \pm 32$ with pure Al (measured from Cm traces)

SPENT FUEL REPROCESSING





- **MINOR ACTINIDES** : *Am - Cm - Np*
(*Mains contributors to long term radiotoxicity, after plutonium*)



- ***SOME FISSION PRODUCTS*** : *I – Cs – Tc*
 - *Significant abundance of long-lived isotope*
 - *More « mobile » in repository conditions*

ACTINIDE PARTITIONING : R&D SUCCESSIVE STEPS...



1. BASIC RESEARCH (*very wide cooperative frame*)

- *innovative : new extractant systems*
- *fundamental : mechanism studies*

*Several hundred
new molecules*

2. DEVELOPMENT

- *process design*
- *hots runs with genuine spent fuel*

Scale 1/10000

- *demonstrative tests*
(integration, representativeness, robustness)

Scale 1/100 à 1/1000